

Self-discharge of all-vanadium liquid flow battery





Overview

What factors contribute to the capacity decay of all-vanadium redox flow batteries?

Learn more. A systematic and comprehensive analysis is conducted on the various factors that contribute to the capacity decay of all-vanadium redox flow batteries, including vanadium ions cross-over, self-discharge reactions, water molecules migration, gas evolution reactions, and vanadium precipitation.

Why are all-vanadium redox flow batteries so popular?

Recently, all-vanadium redox flow batteries (VRFBs) have gained popularity because of their long cycle life, ease of maintenance, and flexible power/capacity configurations. Understanding the process of cell response over time is deemed to be essential for settling the performance-limiting factors.

Do vanadium ions self-discharge?

Sun et al. conducted a detailed study on the self-discharge process, focusing on the self-discharge reactions resulting from the crossover of vanadium ions. They measured the diffusion rates of vanadium ions with different valence states, and the results are presented in Table 3.

What is a vanadium ions cross-diffusion model?

This model enables the simulation of vanadium ions cross-diffusion across the membrane and capacity loss caused by self-discharge reactions in the positive and negative half-cells during long-term charge-discharge cycles of VRFB. By utilizing this model, one can determine when to rebalance the electrolyte to restore battery capacity.



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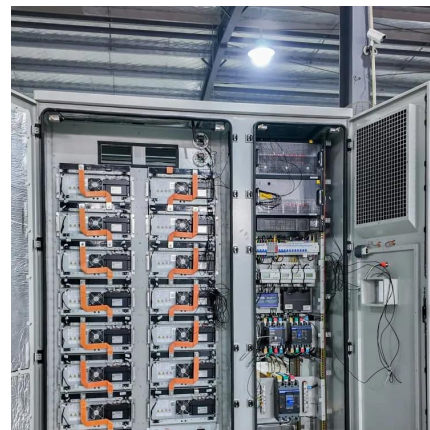
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Motivation and objectives Investigation of self-discharge mechanisms is instrumental in enhancing the efficiency and durability of vanadium flow battery systems (VFBS)

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